GA - General Aptitude

Q1 - Q5 carry one mark each.

Q.No. 1  Rajiv Gandhi Khel Ratna Award was conferred on Mary Kom, a six-time world champion in boxing, recently in a ceremony at the Rashtrapati Bhawan (the President’s official residence) in New Delhi.

(A) with, at
(B) on, in
(C) on, at
(D) to, at

Q.No. 2  Despite a string of poor performances, the chances of K. L. Rahul’s selection in the team are slim.

(A) slim
(B) bright
(C) obvious
(D) uncertain

Q.No. 3  Select the word that fits the analogy:

Cover : Uncover :: Associate :

(A) Unassociate
(B) Inassociate
(C) Misassociate
(D) Dissociate

Q.No. 4  Hit by floods, the kharif (summer sown) crops in various parts of the country have been affected. Officials believe that the loss in production of the kharif crops can be recovered in the output of the rabi (winter sown) crops so that the country can achieve its food-grain production target of 291 million tons in the crop year 2019-20 (July-June). They are hopeful that good rains in July-August will help the soil retain moisture for a longer period, helping winter sown crops such as wheat and pulses during the November-February period.

Which of the following statements can be inferred from the given passage?

(A) Officials declared that the food-grain production target will be met due to good rains.
(B) Officials want the food-grain production target to be met by the November-February period.
(C) Officials feel that the food-grain production target cannot be met due to floods.
(D) Officials hope that the food-grain production target will be met due to a good rabi produce.

Q.No. 5  The difference between the sum of the first 2\(n\) natural numbers and the sum of the first \(n\) odd natural numbers is

(A) \(n^2 - n\)
(B) \(n^2 + n\)
(C) \(2n^2 - n\)
Q6 - Q10 carry two marks each.

Q.No. 6
Repayment rate is the rate at which Reserve Bank of India (RBI) lends commercial banks, and reverse repayment rate is the rate at which RBI borrows money from commercial banks.

Which of the following statements can be inferred from the above passage?

(A) Decrease in repayment rate will increase cost of borrowing and decrease lending by commercial banks.
(B) Increase in repayment rate will decrease cost of borrowing and increase lending by commercial banks.
(C) Increase in repayment rate will decrease cost of borrowing and decrease lending by commercial banks.
(D) Decrease in repayment rate will decrease cost of borrowing and increase lending by commercial banks.

Q.No. 7
P, Q, R, S, T, U, V, and W are seated around a circular table.

I. S is seated opposite to W.
II. U is seated at the second place to the right of R.
III. T is seated at the third place to the left of R.
IV. V is a neighbour of S.

Which of the following must be true?

(A) P is a neighbour of R.
(B) Q is a neighbour of R.
(C) P is not seated opposite to Q.
(D) R is the left neighbour of S.

Q.No. 8
The distance between Delhi and Agra is 233 km. A car P started travelling from Delhi to Agra and another car Q started from Agra to Delhi along the same road 1 hour after the car P started. The two cars crossed each other 75 minutes after the car Q started. Both cars were travelling at constant speed. The speed of car P was 10 km/hr more than the speed of car Q. How many kilometers did car Q travel when the cars crossed each other?

(A) 66.6
(B) 75.2
(C) 88.2
(D) 116.5

Q.No. 9
For a matrix $M = [m_{ij}]$; $i, j = 1, 2, 3, 4$, the diagonal elements are all zero and $m_{ij} = -m_{ji}$.
The minimum number of elements required to fully specify the matrix is_____.

(A) 0
(B) 6
(C) 12
(D) 16

Q.No. 10
The profit shares of two companies P and Q are shown in the figure. If the two companies have invested a fixed and equal amount every year, then the ratio of the total revenue of company P to the total revenue of company Q, during 2013 - 2018 is_____.

![Graph showing profit percentage for companies P and Q from 2013 to 2018.]

(A) 15 : 17
(B) 16 : 17
(C) 17 : 15
(D) 17 : 16

**XL: Life Sciences - P: Chemistry (Compulsory)**

Q1 - Q5 carry one mark each.

Q.No. 1
An aqueous solution contains a mixture of $10^{-8}$ M NaCl and $10^{-8}$ M HCl. Choose the correct statement about this solution.

(A) The solution is a buffer with pH less than 7.00
(B) The solution is a buffer with pH greater than 7.00
(C) The solution is not a buffer but has its pH less than 7.00
(D) The solution is not a buffer but has its pH greater than 7.00

Q.No. 2
The coordination complex which has a distorted octahedral structure is

(Given: Atomic numbers of V: 23; Mn: 25; Ni: 28; Cu: 29)

(A) $[\text{Ni(H}_2\text{O)}_6]^{2+}$
(B) $[\text{Mn(H}_2\text{O)}_6]^{2+}$
(C) $[\text{V(H}_2\text{O)}_6]^{2+}$
(D) $[\text{Cu(H}_2\text{O)}_6]^{2+}$

Q.No. 3
In naphthalene, the value of the integer "n" according to Hückel's rule of aromaticity is______.

Q.No. 4
The azimuthal quantum number ($l$) of an electron in the $d^5$ orbital of a copper atom (atomic number: 29) is__________.

Q.No. 5
The standard enthalpy of reaction (in kJ mol$^{-1}$) for obtaining three moles of H$_2$ (g) from atomic hydrogen in gas phase is__________. (Given: Standard enthalpy of formation of atomic hydrogen in gas phase is 218 kJ mol$^{-1}$)

Q6 - Q15 carry two marks each.
Q.No. 6  The **correct** order of the first ionization energies of He, B, N and O in their corresponding ground state is
(A) He > N > O > B  
(B) O > N > B > He  
(C) He > B > N > O  
(D) N > O > B > He

Q.No. 7  Based on the molecular orbital theory, which one of the following statements with respect to N₂⁺, N₂⁺, O₂ and O₂⁺ is **correct**?
(A) Bond orders of N₂ and O₂ are higher than their corresponding cations.  
(B) Bond energy of N₂⁺ is higher than that of N₂, whereas bond energy of O₂⁺ is lower than that of O₂.  
(C) The unpaired electrons in N₂⁺ and O₂⁺ are present in σ and π* orbitals, respectively.  
(D) The bond in N₂⁺ is shorter than that in N₂, whereas bond in O₂ is shorter than that in O₂⁺.

Q.No. 8  Which one of the following statements is **incorrect** about the diborane molecule?
(A) B-H bond is a 2-centre-2-electron bond (H*: terminal hydrogen).  
(B) BH₃B bond is a 3-centre-2-electron bond (H*: bridged hydrogen).  
(C) The bond angle H₃BH is 122° (H*: terminal hydrogen).  
(D) The B-H bond distance is longer than B-H bond distance (H*: terminal hydrogen, H*: bridged hydrogen).

Q.No. 9  Given below are Newman projections of ethylene glycol and 1,2-difluoroethane about their respective C-C bonds. The most stable conformations (lowest energy) of ethylene glycol and 1,2-difluoroethane are

![Newman projections](image)

(A) I and III respectively.  
(B) I and IV respectively.  
(C) II and III respectively.  
(D) II and IV respectively.

Q.No. 10  In the reaction given below, choose the condition that gives an anti-Markovnikov’s product.

![Reaction](image)

(A) Peroxide / HCl  
(B) Aqueous mercuric acetate treatment  
(C) Diborane addition  
(D) Sulfuric acid addition

Q.No. 11  Which one of the following hexoses will give an osazone that has a different melting point from that of the osazone obtained from D (+) glucose?

(A)
Q.No. 12  A molecule in solution crystallizes into two different crystal forms with rate constants of 0.02 s\(^{-1}\) and 0.13 s\(^{-1}\). If the crystallization is assumed to be under kinetic control, then the half-life (in seconds, rounded off to one decimal place) of the molecule is ________.

Q.No. 13  The standard potential \(E_{\text{cell}}^0\) for a cell reaction given below is +0.7 V. The standard reaction free energy \(\Delta_r G^0\) for this cell is ________ kJ mol\(^{-1}\) (correct up to two decimal places). (Given: Faraday constant, \(F = 96500 \text{ C mol}^{-1}\))

\[
\text{Au}^{3+} (aq) + 3 \text{Ag} (s) \rightarrow \text{Au} (s) + 3 \text{Ag}^{+} (aq)
\]

Q.No. 14  The activation energy \(E_a\) estimated for a reaction from the Arrhenius equation is 21 kJ mol\(^{-1}\). If the frequency factor is assumed to be independent of temperature, then the ratio of the rate constants determined at 298 K and 260 K is ________ (rounded off to two decimal places). (Given: Gas constant, \(R = 8.315 \text{ J K}^{-1} \text{ mol}^{-1}\))

Q.No. 15  At a given pressure, a substance is heated from 2000 K to 2600 K. If the entropy of the substance is 60 J K\(^{-1}\) mol\(^{-1}\), and is assumed to be constant over the given temperature range, then the change in the chemical potential (in kJ mol\(^{-1}\)) of the substance is __________.
Q1 - Q10 carry one mark each.

Q.No. 1 Which one of the following hormones initiates a signaling cascade by directly binding to an intra-cellular receptor?
(A) Insulin
(B) Gonadotropin
(C) Progesterone
(D) Epinephrine

Q.No. 2 Which one of the following bonds is NOT present in ATP?
(A) Phosphoester
(B) Phosphoanhydride
(C) N-Glycosidic
(D) α-Glycosidic

Q.No. 3 The reaction involved in the direct conversion of L-phenylalanine to L-tyrosine is
(A) Hydroxylation
(B) Decarboxylation
(C) Transamination
(D) Reduction

Q.No. 4 The human major histocompatibility complex (MHC) is
(A) Polygenic and monomorphic
(B) Polygenic and polymorphic
(C) Monogenic and polymorphic
(D) Monogenic and monomorphic

Q.No. 5 Har Gobind Khorana and Marshall Nirenberg elucidated the genetic code by using a cell-free protein synthesizing system. It was found that poly(U) and poly(C) result in the synthesis of poly(L-Phe) and poly(L-Pro), respectively. Based on these observations, which one of the following conclusions is correct?
(A) Codon GGG specifies L-Phe and codon AAA specifies L-Pro
(B) Codon CCC specifies L-Phe and codon UUU specifies L-Pro
(C) Codon AAA specifies L-Phe and codon GGG specifies L-Pro
(D) Codon UUU specifies L-Phe and codon CCC specifies L-Pro

Q.No. 6 Binding of an antibody to its cognate antigen does NOT involve
(A) Covalent bonds
(B) Electrostatic forces
(C) Van der Waals forces
(D) Hydrogen bonds

Q.No. 7 A globular protein of molecular weight 50 kDa exists as a mixture of monomers and dimers in solution. The most appropriate technique for the separation of these two forms of the protein is
(A) Thin layer chromatography
(B) Ion exchange chromatography
(C) Gel filtration chromatography
(D) Paper chromatography

Q.No. 8

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17/02/2020 GATE Oالط Official Website, IIT Delhi
gate.iitd.ac.in/XLPaper.php 6/25
Choose the correct order of molecules according to their ability to diffuse across a lipid bilayer.

(A) CO₂ > H₂O > Glucose > RNA
(B) CO₂ > Glucose > H₂O > RNA
(C) RNA > Glucose > CO₂ > H₂O
(D) H₂O > CO₂ > RNA > Glucose

Q.No. 9 When one glucose unit from glycogen gets converted to lactate in the muscle, the net number of ATP molecules produced is ________.

Q.No. 10 Considering that the three pKₐs of histidine are pKₐ₁=1.8, pKₐ₂=9.2 and pKₐ₃=6.0, its isoelectric point will be __________ (rounded off to one decimal place).

Q11 - Q20 carry two marks each.

Q.No. 11 One mole of a native protein upon N-terminal analysis yielded one mole each of Asp and Val. Therefore, the protein in its native state exists as a

(A) Monomer
(B) Homo-dimer
(C) Hetero-dimer
(D) Tetramer

Q.No. 12 The prosthetic groups/cofactors involved in both 1e⁻ and 2e⁻ transfer in the mitochondrial electron transport chain are

(A) NAD and NADP
(B) NAD and FAD
(C) Heme and FMN
(D) Coenzyme Q and FMN

Q.No. 13 Match the items in Group I with the most appropriate items in Group II and choose the correct option.

<table>
<thead>
<tr>
<th>Group I</th>
<th>Group II</th>
</tr>
</thead>
<tbody>
<tr>
<td>P. Integrin</td>
<td>1. Phagocytosis in the neural tissue</td>
</tr>
<tr>
<td>Q. Microglial cell</td>
<td>2. Antigen processing by cross-presentation</td>
</tr>
<tr>
<td>R. TLR-7</td>
<td>3. Single stranded RNA recognition</td>
</tr>
<tr>
<td>S. Dendritic cell</td>
<td>4. Binding of cells to endothelium</td>
</tr>
</tbody>
</table>

(A) P-2, Q-1, R-3, S-4
(B) P-4, Q-1, R-3, S-2
(C) P-1, Q-2, R-3, S-4
(D) P-4, Q-1, R-2, S-3

Q.No. 14 The correct combination of glycosidic linkages present in glycogen is

(A) α 1→4 and α 1→6
**Q.No. 15** Polypeptides are either biosynthesized on the ribosomes using an mRNA template or chemically synthesized by the Merrifield’s solid phase method. The correct directions of peptide synthesis are

(A) C→N direction on the ribosomes and N→C direction on the solid phase
(B) N→C direction on the ribosomes and C→N direction on the solid phase
(C) N→C direction in both cases
(D) C→N direction in both cases

**Q.No. 16** A solution absorbs 20% of the incident light in a cuvette of path length 1.0 cm. The amount of light transmitted by the same solution in a cuvette of 3.0 cm path length is _______ % (rounded off to one decimal place).

**Q.No. 17** The second pKa of phosphoric acid is 6.8. The ratio of Na₃H₂PO₄ to NaH₂PO₄ required to obtain a buffer of pH 7.0 is ________ (rounded off to two decimal places).

**Q.No. 18** A PCR in a 100 μL reaction volume, containing two primers at a concentration of 0.2 μM each, is set up to amplify a 250 base pair DNA fragment. Consider the average molecular weight of one base pair as 660 Da. If the primers are fully consumed by the end of the reaction, the amount of the final PCR product formed is ________ μg (rounded off to one decimal place).

**Q.No. 19** An enzyme obeying Michaelis-Menten kinetics shows a reaction velocity (v) of 10 μmol/min when the substrate concentration [S] equals its Kₘ. The maximal velocity Vₘₐₓ for this enzyme is ______ μmol/min (correct to integer number). (Kₘ is Michaelis-Menten constant)

**Q.No. 20** The enzyme glucose isomerase catalyzes the inter-conversion of glucose and fructose as shown.

\[
\text{Glucose} \rightleftharpoons \text{Fructose}
\]

The ΔG° for this reaction is zero kcal/mol. After adding glucose isomerase to a 0.12 M glucose solution and allowing the reaction to attain equilibrium, the final concentration of fructose in the reaction mixture will be ________ mM.

**XL: Life Sciences - R: Botany**

**Q1 - Q10 carry one mark each.**

**Q.No. 1** Indefinite stamen is a characteristic feature of which of the following plant families?

(A) Malvaceae
(B) Apocynaceae
(C) Poaceae
(D) Brassicaceae
In natural condition, which of the following plants **DOES NOT** exhibit anomalous secondary growth?

(A) Rice  
(B) Aloe  
(C) Yucca  
(D) Dracaena

**Q.No. 3** In a typical angiosperm under natural condition, primary meristems are usually established during

(A) Gametogenesis  
(B) Embryogenesis  
(C) Vegetative phase development  
(D) Secondary growth

**Q.No. 4** 2-Methoxy-3, 6-dichlorobenzoic acid belongs to which class of plant growth regulators?

(A) Synthetic auxin  
(B) Synthetic cytokinin  
(C) Strigolactone  
(D) Brassinosteroid

**Q.No. 5** In a typical green plant, the first stable product of Calvin cycle is

(A) Oxaloacetic acid  
(B) Succinic acid  
(C) Maleic acid  
(D) 3-Phosphoglyceric acid

**Q.No. 6** Among the following, which best describes an organism that lives at the expense of other organisms, harmful but usually not killing?

(A) Predator  
(B) Symbiotic  
(C) Prey  
(D) Parasite

**Q.No. 7** The oleo-gum resin asafoetida (hing) is obtained from the cut surface of

(A) Stem  
(B) Root  
(C) Leaf  
(D) Fruit

**Q.No. 8** ‘Bakanae’ disease or ‘foolish seedling’ disease is caused by

(A) Fungus  
(B) Bacterium  
(C) Virus  
(D) Mycoplasma

**Q.No. 9** Which of the following chemicals is used for doubling of chromosome numbers during production of ‘doubled haploids’ in crop plants?

(A) Hygromycin  
(B) Kanamycin  
(C) Colchicine  
(D) Glufosinate
Q.No. 10
An mRNA of a nuclear encoded plant gene, DSH20 has an ORF of 1353 nucleotides. Provided that average molecular weight of amino acid is 110 Dalton (Da), calculated molecular weight of DSH20 protein in kDa (round off to 1 decimal place) is ____.

Q. No. 11 - Q20 carry two marks each.

Group I, Group II and Group III represent enzyme, product of the enzymatic reaction, and metabolic process, respectively.

<table>
<thead>
<tr>
<th>Group I</th>
<th>Group II</th>
<th>Group III</th>
</tr>
</thead>
<tbody>
<tr>
<td>P. Hexokinase</td>
<td>i. Malate</td>
<td>1. Glycolysis</td>
</tr>
<tr>
<td>Q. Fumarase</td>
<td>ii. Glucose 6-P</td>
<td>2. Photorespiration</td>
</tr>
<tr>
<td>R. PEP Carboxylase</td>
<td>iii. Hydrogen peroxide</td>
<td>3. TCA cycle</td>
</tr>
<tr>
<td>S. Glycolate oxidase</td>
<td>iv. Cinnamic acid</td>
<td>4. Photosynthesis</td>
</tr>
<tr>
<td></td>
<td>v. Oxaloacetic acid</td>
<td></td>
</tr>
</tbody>
</table>

The **CORRECT** combination for **Group I, Group II** and **Group III** is
(A) P-ii, Q-iv, R-v, S-iii
(B) P-ii, Q-i, R-iv, S-iii
(C) P-iii, Q-i, R-iii, S-iv
(D) P-iii, Q-i, R-iv, S-ii

Q.No. 12
Match the following in **CORRECT** combination between **Group I** and **Group II** with reference to the agents that interfere with oxidative phosphorylation.

<table>
<thead>
<tr>
<th>Group I</th>
<th>Group II</th>
</tr>
</thead>
<tbody>
<tr>
<td>P. Cyanide</td>
<td>i. Blocks electron transfer from cyt b to cyt c1</td>
</tr>
<tr>
<td>Q. Antimycin A</td>
<td>ii. Inhibits F1</td>
</tr>
<tr>
<td>R. Aurovertin</td>
<td>iii. Uncoupling of phosphorylation from electron transfer</td>
</tr>
<tr>
<td>S. 2,4-Dinitrophenol</td>
<td>iv. Inhibits cytochrome oxidase</td>
</tr>
<tr>
<td></td>
<td>v. Inhibits K⁺ ionophore</td>
</tr>
</tbody>
</table>

(A) P-iv, Q-i, R-ii, S-iii
(B) P-v, Q-i, R-iii, S-iv
(C) P-iv, Q-iii, R-ii, S-v
(D) P-v, Q-ii, R-iii, S-iv

Q.No. 13
In relation to Agrobacterium mediated genetic engineering in plants, match the following in \textbf{CORRECT} combination

<table>
<thead>
<tr>
<th>Gene name</th>
<th>Function</th>
</tr>
</thead>
<tbody>
<tr>
<td>P. \textit{virA}</td>
<td>i. Acetosyringone receptor</td>
</tr>
<tr>
<td>Q. \textit{virB}</td>
<td>ii. Conjugal tube formation</td>
</tr>
<tr>
<td>R. \textit{virD}</td>
<td>iii. Topoisomerase</td>
</tr>
<tr>
<td>S. \textit{virG}</td>
<td>iv. Inducer of all \textit{vir} operons</td>
</tr>
<tr>
<td></td>
<td>v. Octopine synthesis</td>
</tr>
</tbody>
</table>

(A) P-iv, Q-iii, R-ii, S-v  
(B) P-ii, Q-i, R-iii, S-v  
(C) P-i, Q-ii, R-iii, S-iv  
(D) P-iii, Q-i, R-ii, S-iv

\textbf{Q.No. 14} Match the plant part (\textbf{Group I}) with the product obtained (\textbf{Group II}) and the representative plant species (\textbf{Group III}) in \textbf{CORRECT} combination

<table>
<thead>
<tr>
<th>Group I</th>
<th>Group II</th>
<th>Group III</th>
</tr>
</thead>
<tbody>
<tr>
<td>P. Bark</td>
<td>i. Tannins</td>
<td>1. \textit{Papaver somniferum}</td>
</tr>
<tr>
<td>Q. Leaf</td>
<td>ii. Saffron</td>
<td>2. \textit{Camellia sinensis}</td>
</tr>
<tr>
<td>R. Flower</td>
<td>iii. Codeine</td>
<td>3. \textit{Cinnamomum zeylanicum}</td>
</tr>
<tr>
<td>S. Fruit</td>
<td>iv. Aromatic oil</td>
<td>4. \textit{Crocus sativus}</td>
</tr>
</tbody>
</table>

(A) P-ii, Q-3, R-4, S-iii-4  
(B) P-ii, Q-4, R-2, S-iii-4  
(C) P-ii, Q-3, R-4, S-iii-4  
(D) P-iv, Q-2, R-ii-4, S-iii-1

\textbf{Q.No. 15} Select the \textbf{CORRECT} combination by matching the disease, causal organism and the affected plant.

<table>
<thead>
<tr>
<th>Disease</th>
<th>Causal organism</th>
<th>Affected plant</th>
</tr>
</thead>
<tbody>
<tr>
<td>P. Stem rust</td>
<td>i. \textit{Cercospora personata}</td>
<td>1. Wheat</td>
</tr>
<tr>
<td>Q. Wart disease</td>
<td>ii. \textit{Plasmopara viticola}</td>
<td>2. Ground nut</td>
</tr>
<tr>
<td>R. Tikka/leaf spot</td>
<td>iii. \textit{Synchytrium endobioticum}</td>
<td>3. Potato</td>
</tr>
</tbody>
</table>

(A) P-iv-1, Q-iii-3, R-i-2, S-ii-4
Q.No. 16  Match the following alkaloids with their uses and source plants in **CORRECT** combination

<table>
<thead>
<tr>
<th>Alkaloid</th>
<th>Use</th>
<th>Source plant</th>
</tr>
</thead>
<tbody>
<tr>
<td>P. Morphine</td>
<td>i. Anti-cancer</td>
<td>1. <em>Cinchona officinalis</em></td>
</tr>
<tr>
<td>Q. Quinine</td>
<td>ii. Analgesic</td>
<td>2. <em>Catharanthus roseus</em></td>
</tr>
<tr>
<td>R. Atropine</td>
<td>iii. Anti-cholinergic</td>
<td>3. <em>Papaver somniferum</em></td>
</tr>
<tr>
<td>S. Vinblastine</td>
<td>iv. Anti-malarial</td>
<td>4. <em>Hyoscyamus niger</em></td>
</tr>
</tbody>
</table>

(A) P-ii-3, Q-iv-1, R-iii-4, S-ii-2  
(B) P-ii-1, Q-iii-3, R-iv-4, S-iii-2  
(C) P-ii-2, Q-iv-1, R-i-4, S-iii-3  
(D) P-iii-4, Q-ii-1, R-iv-3, S-i-2

Q.No. 17  Match the following ecological terms with their appropriate definitions

<table>
<thead>
<tr>
<th>Term</th>
<th>Definition</th>
</tr>
</thead>
<tbody>
<tr>
<td>P. Niche</td>
<td>i. Position of a species in food chain</td>
</tr>
<tr>
<td>Q. Biotas</td>
<td>ii. Place of a living organism in the biotic environment and its relations to food and enemies</td>
</tr>
<tr>
<td>R. Trophic level</td>
<td>iii. Physical environment of an organism</td>
</tr>
<tr>
<td>S. Habitat</td>
<td>iv. Totality of organisms (flora and fauna) in a given place or region</td>
</tr>
</tbody>
</table>

(A) P-i, Q-ii, R-iv, S-iii  
(B) P-ii, Q-iv, R-i, S-iii  
(C) P-iv, Q-iii, R-i, S-ii  
(D) P-iii, Q-i, R-ii, S-iv

Q.No. 18  Arrange the following ‘water reservoirs of earth’ in decreasing order of water volume

P- Streams  
Q- Groundwater  
R- Glaciers  
S- Lakes and inland seas

(A) R-Q-S-P  
(B) P-Q-R-S  
(C) S-P-R-Q
Q.No. 19  
Selection markers and the corresponding genes used in plant genetic engineering are given below:

<table>
<thead>
<tr>
<th>Selection Marker</th>
<th>Gene</th>
</tr>
</thead>
<tbody>
<tr>
<td>P. Kanamycin</td>
<td>i. hptIV</td>
</tr>
<tr>
<td>Q. Hygromycin</td>
<td>ii. bar</td>
</tr>
<tr>
<td>R. Bialaphos</td>
<td>iii. pmi</td>
</tr>
<tr>
<td>S. Mannose</td>
<td>iv. nptII</td>
</tr>
</tbody>
</table>

Choose the CORRECT combination:
(A) P-ii, Q-i, R-iv, S-iii  
(B) P-iv, Q-ii, R-i, S-iii  
(C) P-iv, Q-i, R-ii, S-iii  
(D) P-iii, Q-iv, R-ii, S-i  

Q.No. 20  
A double homozygous mutant develops green and wrinkled seeds. When it was crossed with a true-breeding plant having yellow and round seeds, all the F1 plants developed yellow and round seeds. After self-fertilization of F1, the calculated percentage probability of plants with green and wrinkled seeds in the F2 population (round off to 2 decimal places) is _________

XL: Life Sciences - S: Microbiology
Q1 - Q10 carry one mark each.

Q.No. 1  
The technique of microbial “pure culture” was pioneered by
(A) Edward Jenner  
(B) Louis Pasteur  
(C) Robert Hooke  
(D) Robert Koch  

Q.No. 2  
The antibacterial trimethoprim is an inhibitor of
(A) dihydrofolate reductase  
(B) dihydropterin reductase  
(C) Δ^5,Δ^10-methenyl tetrahydrofolate synthetase  
(D) serine hydroxymethyl transferase  

Q.No. 3  
Choose the correct taxonomical hierarchy among the following:
(A) Species, Genus, Family, Order, Class, Phylum, Domain  
(B) Species, Genus, Order, Class, Family, Phylum, Domain  
(C) Species, Genus, Order, Family, Class, Phylum, Domain  
(D) Species, Genus, Family, Class, Order, Phylum, Domain  

Q.No. 4  
Shifting a Saccharomyces cerevisiae culture from fermentative to aerobic respiratory mode will
(A) decrease carbon dioxide production  
(B) increase alcohol production
Q. No. 5 Which one of the following diseases is treated by a neuraminidase inhibitor?
(A) Chickenpox  
(B) Polio  
(C) Influenza  
(D) Japanese encephalitis

Q. No. 6 Which one of the following does NOT provide three-dimensional images?
(A) Atomic force microscopy  
(B) Confocal scanning laser microscopy  
(C) Differential interference contrast microscopy  
(D) Phase-contrast microscopy

Q. No. 7 Which one of the following will increase the resolution of a light microscope?
(A) Decreasing the numerical aperture of the objective lens  
(B) Using an objective lens with a longer working distance  
(C) Using a medium of higher refractive index  
(D) Increasing the wavelength of light

Q. No. 8 Which one of the following conditions favors maximum expression of lac operon genes in E. coli?
(A) Glucose-low, lactose-low, cAMP-high  
(B) Glucose-high, lactose-low, cAMP-high  
(C) Glucose-low, lactose-high, cAMP-high  
(D) Glucose-high, lactose-high, cAMP-low

Q. No. 9 Match the cellular organelle in Group I with its function in Group II

<table>
<thead>
<tr>
<th>Group I</th>
<th>Group II</th>
</tr>
</thead>
<tbody>
<tr>
<td>P. Golgi apparatus</td>
<td>1. Lipid degradation</td>
</tr>
<tr>
<td>Q. Nucleolus</td>
<td>2. Protein degradation</td>
</tr>
<tr>
<td>R. Peroxisome</td>
<td>3. Protein sorting</td>
</tr>
<tr>
<td>S. Proteasome</td>
<td>4. Ribosomal RNA synthesis</td>
</tr>
</tbody>
</table>

| (A) P-3, Q-2, R-1, S-4  
| (B) P-3, Q-4, R-1, S-2  
| (C) P-1, Q-2, R-4, S-3  
| (D) P-3, Q-1, R-4, S-2 |

Q. No. 10 A 250 μl of bacteriophage stock containing $8 \times 10^8$ phages/ml is added to 500 μl of E. coli culture containing $4 \times 10^8$ cells/ml. The multiplicity of infection is _____.

Q11 - Q20 carry two marks each.

Q. No. 11 Digestion of an immunoglobulin G (IgG) molecule with pepsin will NOT
(A) generate a bivalent antigen binding fragment  
(B) generate monovalent antigen binding fragments  
(C)
destroy the complement binding site

Q.No. 12 Match the process involved in nitrogen or sulfur cycle in Group I with the corresponding microbe in Group II.

<table>
<thead>
<tr>
<th>Group I</th>
<th>Group II</th>
</tr>
</thead>
<tbody>
<tr>
<td>P. Denitration</td>
<td>1. Azotobacter</td>
</tr>
<tr>
<td>Q. Nitrogen fixation by free-living microbe</td>
<td>2. Beggiatoa</td>
</tr>
<tr>
<td>R. Oxidation of H₂S to sulfur</td>
<td>3. Pseudomonas</td>
</tr>
<tr>
<td>S. Nitrogen fixation by a symbiotic microbe</td>
<td>4. Rhizobium</td>
</tr>
</tbody>
</table>

(A) P-2, Q-3, R-4, S-1
(B) P-2, Q-1, R-3, S-4
(C) P-3, Q-4, R-1, S-2
(D) P-3, Q-1, R-2, S-4

Q.No. 13 Determine the correctness or otherwise of the following Assertion [a] and the Reason [r].

Assertion [a]: Diphtheria exotoxin is an example of A-B type toxin.

Reason [r]: The A component of the toxin is released from the host cell while the B component inhibits protein synthesis and kills the host cell.

(A) Both [a] and [r] are true and [r] is the correct reason for [a]
(B) Both [a] and [r] are true but [r] is not the correct reason for [a]
(C) Both [a] and [r] are false
(D) [a] is true but [r] is false

Q.No. 14 Which one of the following statements about control of microbial growth is NOT correct?

(A) Nonionizing radiation leads to thymine dimers formation in DNA
(B) Spirochetes and mycoplasma can pass through membrane filters (0.22-0.45 μm)
(C) Use of high concentration of salts and sugars to preserve food is a chemical method of microbial control
(D) Thermophilic bacteria can survive pasteurization

Q.No. 15 An example of a differential and selective medium in which colonies of Gram-negative bacteria produce large amounts of acidic products and appear green with a metallic sheen is

(A) Blood agar
(B) EMB agar
(C) MacConkey agar
(D) Mannitol salt agar

Q.No. 16 Which one of the following is an example of substrate level phosphorylation?

(A) Glucose to Glucose 6-phosphate
(B) Fructose 6-phosphate to Fructose 1,6-bisphosphate
Q.No. 17 A bacterial culture containing $3 \times 10^5$ live cells was exposed to a newly developed sterilizing agent. After 30 minutes of exposure, 3 live cells remained in culture. The decimal reduction time (in minutes) for the new agent is _____.

Q.No. 18 A bacterial culture has a generation time of 34 minutes. The time taken (in minutes, rounded off to two decimal places) for the OD$_{550}$ of this exponentially growing culture to increase from 0.25 to 0.85 is _____.

Assume that OD$_{550}$ has a linear relationship with the cell density.

Q.No. 19 A 100 µl aliquot ($10^{-4}$ dilution) of the bacterial culture plated on the nutrient agar gave 4 colonies. The bacterial stock concentration (in million cells/ml, rounded off to one decimal place) is _____.

Q.No. 20 A continuous bacterial culture carried out in a chemostat is set to a flow rate of 40 ml/hr. The culture volume is equivalent to that of a cubical container having 10 cm sides. The dilution rate (in hr$^{-1}$, rounded off to two decimal places) of this system is _____.

XL: Life Sciences - T: Zoology

Q1 - Q10 carry one mark each.

Q.No. 1 Which ONE of following leucocytes is phagocytic and has clear cytoplasm?
(A) Eosinophil
(B) Monocyte
(C) T$_H$-lymphocyte
(D) Basophil

Q.No. 2 Which ONE of the following techniques can be used for detecting the subcellular localization of serotonin receptor in intact cells?
(A) Immunelectron microscopy
(B) SDS-PAGE
(C) Fluorescence in-situ hybridization
(D) Differential centrifugation

Q.No. 3 Which ONE of the following is NOT a site for in situ conservation?
(A) Biosphere reserve
(B) Wild life sanctuary
(C) Zoological garden
(D) Biodiversity hotspot

Q.No. 4 Which ONE of the following is the precursor molecule for corticosteroids?
(A) Androgen
(B) Estrogen
(C) Pregnenolone
(D) Mineralocorticoids
Q.No. 5  Transitional epithelia is found in which ONE of the following organs?
(A) Liver
(B) Lung
(C) Brain
(D) Urinary bladder

Q.No. 6  Visual signal transduction cascade is activated by rhodopsin and involves degradation rather than synthesis of which ONE of the following second messenger molecules?
(A) cAMP
(B) IP3
(C) cGMP
(D) DAG

Q.No. 7  The genomes of both human and Drosophila code for an amylase that acts on the same substrate. However, the sequence of nucleotides in the genes encoding the two is dissimilar. This is an example of which ONE of the following types of evolution?
(A) Neutral
(B) Directional
(C) Convergent
(D) Divergent

Q.No. 8  “Round dance” is performed by forager bees to indicate the distance between a food source and their colony. Which ONE of the following best represents this distance?
(A) 45 meters
(B) 450 meters
(C) 1000 meters
(D) More than 2000 meters

Q.No. 9  Which ONE of the following phyla have choanocytes?
(A) Ctenophora
(B) Nematoda
(C) Cnidaria
(D) Porifera

Q.No. 10  Which ONE of the following glial cells is NOT derived from the ectoderm?
(A) Astrocytes
(B) Microglial cells
(C) Oligodendrocytes
(D) Ependyma

Q11 - Q20 carry two marks each.

Q.No. 11  Tarantulas and mosquitoes both belong to the phylum Arthropoda. Which ONE of the following represents the correct number of legs in them respectively?
(A) 6 and 6
(B) 6 and 8
(C) 8 and 8
(D) 8 and 6
Match the following subcellular organelles in Column I with associated functions in Column II

<table>
<thead>
<tr>
<th>Column I</th>
<th>Column II</th>
</tr>
</thead>
<tbody>
<tr>
<td>P. Nucleolus</td>
<td>(i) Glycoprotein biosynthesis</td>
</tr>
<tr>
<td>Q. Peroxisomes</td>
<td>(ii) Oxidation of fatty acids and amino acids</td>
</tr>
<tr>
<td>R. Endoplasmic reticulum</td>
<td>(iii) Protein trafficking</td>
</tr>
<tr>
<td>S. Golgi bodies</td>
<td>(iv) Ribosome biogenesis</td>
</tr>
</tbody>
</table>

(A) P-(iii), Q-(ii), R-(i), S-(iv)
(B) P-(i), Q-(ii), R-(iii), S-(iv)
(C) P-(iv), Q-(ii), R-(i), S-(iii)
(D) P-(ii), Q-(iii), R-(i), S-(iv)

Q.No. 13 Match the following genetic disorders in Column I with associated typical chromosomal changes mentioned in Column II

<table>
<thead>
<tr>
<th>Column I</th>
<th>Column II</th>
</tr>
</thead>
<tbody>
<tr>
<td>P. Klinefelter syndrome</td>
<td>(i) 45,XO</td>
</tr>
<tr>
<td>Q. Down syndrome</td>
<td>(ii) 5p minus</td>
</tr>
<tr>
<td>R. Turner syndrome</td>
<td>(iii) 47,XXY</td>
</tr>
<tr>
<td>S. Cri du chat syndrome</td>
<td>(iv) Trisomy 21</td>
</tr>
</tbody>
</table>

(A) P-(iv), Q-(iii), R-(ii), S-(i)
(B) P-(iv), Q-(ii), R-(i), S-(iii)
(C) P-(iii), Q-(iv), R-(ii), S-(i)
(D) P-(iii), Q-(iv), R-(i), S-(ii)

Q.No. 14 Match the following components listed in Column I with their respective organs in Column II

<table>
<thead>
<tr>
<th>Column I</th>
<th>Column II</th>
</tr>
</thead>
<tbody>
<tr>
<td>P. Endolymph</td>
<td>(i) Testes</td>
</tr>
<tr>
<td>Q. Vitreous humour</td>
<td>(ii) Ear</td>
</tr>
<tr>
<td>R. Vas deferens</td>
<td>(iii) Ovary</td>
</tr>
<tr>
<td>S. Corpus luteum</td>
<td>(iv) Eye</td>
</tr>
</tbody>
</table>

(A) P-(ii), Q-(iv), R-(i), S-(iii)
(B) P-(ii), Q-(i), R-(iv), S-(iii)
(C) P-(iii), Q-(iv), R-(i), S-(ii)
(D) P-(iii), Q-(iv), R-(ii), S-(i)

Q.No. 15
Match the following digestive enzymes in **Column I** with their respective functions in **Column II**.

<table>
<thead>
<tr>
<th>Column I</th>
<th>Column II</th>
</tr>
</thead>
<tbody>
<tr>
<td>P. Erespin</td>
<td>(i) converts proteins to peptides</td>
</tr>
<tr>
<td>Q. Steapsin</td>
<td>(ii) activates trypsinogen to trypsin</td>
</tr>
<tr>
<td>R. Pepsin</td>
<td>(iii) converts fat into fatty acid and glycerol</td>
</tr>
<tr>
<td>S. Enterokinase</td>
<td>(iv) converts polypeptides to amino acids</td>
</tr>
</tbody>
</table>

(A) P-(iv), Q-(iii), R-(ii), S-(i)
(B) P-(iv), Q-(ii), R-(i), S-(ii)
(C) P-(iii), Q-(iv), R-(i), S-(ii)
(D) P-(iii), Q-(iv), R-(ii), S-(i)

**Q.No. 16** Which ONE of the following graphs represents the relationship between ventricular end-diastolic volume and cardiac output in a healthy adult individual at rest (solid line) and upon exercise (dotted line)?

(A) [Graph A]
(B) [Graph B]
(C) [Graph C]
(D) [Graph D]

**Q.No. 17**
Match the household insect vectors in **Column I** with their associated diseases in **Column II**.

<table>
<thead>
<tr>
<th>Column I</th>
<th>Column II</th>
</tr>
</thead>
<tbody>
<tr>
<td>P. Kissing bug (Hemiptera)</td>
<td>(i) Bubonic plague</td>
</tr>
<tr>
<td>Q. Sand fly (Diptera)</td>
<td>(ii) Tularemia</td>
</tr>
<tr>
<td>R. Deer fly (Diptera)</td>
<td>(iii) Chagas disease</td>
</tr>
<tr>
<td>S. Oriental rat flea (Siphonoptera)</td>
<td>(iv) Kala azar</td>
</tr>
</tbody>
</table>

(A) P-(iv), Q-(iii), R-(ii), S-(i)
(B) P-(iii), Q-(ii), R-(i), S-(iv)
(C) P-(i), Q-(iv), R-(iii), S-(ii)
(D) P-(iii), Q-(iv), R-(ii), S-(i)

Q.No. 18 Match the proteins in **Column I** with the organs in which they are maximally expressed in **Column II**.

<table>
<thead>
<tr>
<th>Column I</th>
<th>Column II</th>
</tr>
</thead>
<tbody>
<tr>
<td>P. Keratin</td>
<td>(i) Liver</td>
</tr>
<tr>
<td>Q. Surfactants</td>
<td>(ii) Pancreas</td>
</tr>
<tr>
<td>R. Pro-carboxypeptidase</td>
<td>(iii) Lung</td>
</tr>
<tr>
<td>S. Albumin</td>
<td>(iv) Skin</td>
</tr>
</tbody>
</table>

(A) P-(iv), Q-(i), R-(iii), S-(ii)
(B) P-(iii), Q-(iv), R-(ii), S-(i)
(C) P-(iv), Q-(iii), R-(ii), S-(i)
(D) P-(i), Q-(ii), R-(iii), S-(iv)

Q.No. 19 The graph below shows the activity of enzyme pepsin in the presence of inhibitors aliphatic alcohols (P) or N-acetyl-L-phenylalanine (Q). Which ONE of the following represents the nature of inhibition by P and Q, respectively?

![Graph of enzyme activity](image)

(A) Non-competitive and competitive
(B) Competitive and non-competitive
(C) Non-competitive and uncompetitive
(D) Competitive and uncompetitive

Q.No. 20
In Drosophila, the red eye phenotype (W) is dominant over the recessive white eye mutant (w). In a mixed population of red and white eye flies of 10,000 individuals, 3,600 flies were white eyed. The percentage of the heterozygous red eye flies in this population is ________________.

XL: Life Sciences - U: Food Technology

Q1 - Q10 carry one mark each.

Q.No. 1 The enzyme majorly involved in postmortem degradation of muscle proteins is
(A) Trypsin  
(B) Calpin  
(C) Transglutaminase  
(D) Pepsin

Q.No. 2 Which of the following is the correct pair of essential fatty acids?
(A) Oleic acid and Lenoleic acid  
(B) Lenoleic acid and Linolenic acid  
(C) Linolenic acid and Lauric acid  
(D) Linolenic acid and Oleic acid

Q.No. 3 Nisin A is produced by
(A) Aspergillus niger  
(B) Acetobacter acetii  
(C) Lactobacillus lactis  
(D) Clostridium perfringens

Q.No. 4 Which of the following bacteria will stain purple color after Gram staining?
(A) Bacillus subtilis  
(B) Escherichia coli  
(C) Pseudomonas aeruginaosa  
(D) Yersinia pestis

Q.No. 5 The enzyme system used for removal of glucose from egg white prior to its drying consists of
(A) Glucose oxidase and Catalase  
(B) Glucosidase and Glucoisomerase  
(C) Glucoisomerase and Catalase  
(D) Glucoamylase and Glucose oxidase

Q.No. 6 The INCORRECT pair of food borne illness and its causative microorganism is
(A) Brucellosis – Brucella Sp.  
(B) Peptic ulcers – Bacillus subtilis  
(C) Bubonic plague - Yersinia pestis  
(D) Q fever – Coxiella burnetii

Q.No. 7 Which of the following is commonly used as a preservative in the tomato sauce?
(A) Sodium sulphite  
(B) Potassium sorbate  
(C) Potassium sulphite  
(D) Sodium benzoate

Q.No. 8 A fluid with flow behaviour index less than one (n < 1) is
(A) Dilatant
Q.No. 9
The velocity of 2.2 µm diameter fat particles inside a centrifuge, running at 6000 rpm and 20°C, is 0.25 mm s⁻¹. The velocity of 1.5 µm diameter fat particles inside the same centrifuge running at 7500 rpm and same temperature (round off to 2 decimal places) will be _______ mm s⁻¹.

Q.No. 10
The initial population of a bacterial strain increases from $1 \times 10^4$ cells per mL to $1 \times 10^6$ cells per mL in 120 minutes. The generation time for this strain (round off to 2 decimal places) is _______ minutes.

Q11 - Q20 carry two marks each.

Q.No. 11
Match the protein in Column I with its food source in Column II.

<table>
<thead>
<tr>
<th>Column I</th>
<th>Column II</th>
</tr>
</thead>
<tbody>
<tr>
<td>P. Zein</td>
<td>1. Soybean</td>
</tr>
<tr>
<td>Q. Gluten</td>
<td>2. Maize</td>
</tr>
<tr>
<td>R. Glycinin</td>
<td>3. Egg</td>
</tr>
<tr>
<td>S. Ovalbumin</td>
<td>4. Wheat</td>
</tr>
</tbody>
</table>

(A) P-4, Q-1, R-2, S-3
(B) P-4, Q-3, R-1, S-2
(C) P-2, Q-3, R-1, S-4
(D) P-2, Q-4, R-1, S-3

Q.No. 12
Match the carbohydrate in Column I with corresponding enzyme used for its hydrolysis in Column II.

<table>
<thead>
<tr>
<th>Column I</th>
<th>Column II</th>
</tr>
</thead>
<tbody>
<tr>
<td>P. Pectin</td>
<td>1. Xylanase</td>
</tr>
<tr>
<td>Q. Lactose</td>
<td>2. β-galactosidase</td>
</tr>
<tr>
<td>R. Hemicellulose</td>
<td>3. Polysaccharidase</td>
</tr>
<tr>
<td>S. Inulin</td>
<td>4. β-fructofuranosidase</td>
</tr>
</tbody>
</table>

(A) P-3, Q-2, R-1, S-4
(B) P-2, Q-4, R-1, S-3
(C) P-1, Q-2, R-3, S-4
(D) P-4, Q-3, R-1, S-2

Q.No. 13
Match the edible oil refining stage in Column I with its purpose in Column II.

<table>
<thead>
<tr>
<th>Column I</th>
<th>Column II</th>
</tr>
</thead>
<tbody>
<tr>
<td>P. Degumming</td>
<td>1. Separation of triglycerides</td>
</tr>
<tr>
<td>Q. Neutralization</td>
<td>2. Removal of pigments</td>
</tr>
<tr>
<td>R. Bleaching</td>
<td>3. Removal of phosphatides</td>
</tr>
<tr>
<td>S. Winterization</td>
<td>4. Removal of free fatty acids</td>
</tr>
</tbody>
</table>

(A) P-3, Q-1, R-2, S-4
Q.No. 14  Match the food material in Column I with its related term in Column II.

<table>
<thead>
<tr>
<th>Column I</th>
<th>Column II</th>
</tr>
</thead>
<tbody>
<tr>
<td>P. Coffee</td>
<td>1. Wort</td>
</tr>
<tr>
<td>Q. Cocoa</td>
<td>2. Must</td>
</tr>
<tr>
<td>R. Beer</td>
<td>3. <em>Arabica</em></td>
</tr>
<tr>
<td>S. Wine</td>
<td>4. <em>Theobroma</em></td>
</tr>
</tbody>
</table>

(A) P-4, Q-2, R-1, S-3  
(B) P-3, Q-4, R-1, S-2  
(C) P-3, Q-4, R-2, S-1  
(D) P-1, Q-3, R-4, S-2

Q.No. 15  Match the component/system in Column I with the peeling method for fruits and vegetables in Column II.

<table>
<thead>
<tr>
<th>Column I</th>
<th>Column II</th>
</tr>
</thead>
<tbody>
<tr>
<td>P. Lye solution</td>
<td>1. Flash peeling</td>
</tr>
<tr>
<td>Q. Carborundum rollers</td>
<td>2. Flame peeling</td>
</tr>
<tr>
<td>R. Pressure vessel</td>
<td>3. Abrasion peeling</td>
</tr>
<tr>
<td>S. Conveyor belt</td>
<td>4. Caustic peeling</td>
</tr>
</tbody>
</table>

(A) P-4, Q-3, R-2, S-1  
(B) P-3, Q-4, R-1, S-2  
(C) P-4, Q-3, R-1, S-2  
(D) P-3, Q-4, R-2, S-1

Q.No. 16  Which among the given options correctly explains the nature of the microbial culture represented by lines 1, 2 and 3 in the following figure?

<table>
<thead>
<tr>
<th>(A)</th>
</tr>
</thead>
<tbody>
<tr>
<td>1. Germination of spores</td>
</tr>
<tr>
<td>2. Homogeneous population</td>
</tr>
<tr>
<td>3. Mixed population of spores and vegetative cells</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>(B)</th>
</tr>
</thead>
<tbody>
<tr>
<td>1. Homogeneous population</td>
</tr>
<tr>
<td>2. Mixed population of heat sensitive and heat resistant microbes</td>
</tr>
<tr>
<td>3. Germination of spores</td>
</tr>
</tbody>
</table>
Q.No. 17
Match the equation/law in Column I with its application in Column II.

<table>
<thead>
<tr>
<th>Column I</th>
<th>Column II</th>
</tr>
</thead>
<tbody>
<tr>
<td>P. Plank’s equation</td>
<td>1. Terminal velocity</td>
</tr>
<tr>
<td>Q. Arrhenius equation</td>
<td>2. Freezing time</td>
</tr>
<tr>
<td>R. Guggenheim-Anderson-de Boer equation</td>
<td>3. Activation energy</td>
</tr>
<tr>
<td>S. Stoke’s law</td>
<td>4. Monolayer moisture content</td>
</tr>
</tbody>
</table>

(A) P-1, Q-3, R-4, S-2  
(B) P-2, Q-3, R-1, S-4  
(C) P-2, Q-3, R-4, S-1  
(D) P-4, Q-3, R-1, S-2

Q.No. 18
Match the absorber used in modified atmosphere packaging and storage in Column I with the scavenger in Column II.

<table>
<thead>
<tr>
<th>Column I</th>
<th>Column II</th>
</tr>
</thead>
<tbody>
<tr>
<td>P. Oxygen absorber</td>
<td>1. Calcium chloride</td>
</tr>
<tr>
<td>Q. Carbon dioxide absorber</td>
<td>2. Magnesium oxide</td>
</tr>
<tr>
<td>R. Ethylene absorber</td>
<td>3. Ferric oxide</td>
</tr>
<tr>
<td>S. Moisture absorber</td>
<td>4. Potassium permanganate</td>
</tr>
</tbody>
</table>

(A) P-3, Q-2, R-4, S-1  
(B) P-1, Q-2, R-4, S-3  
(C) P-2, Q-3, R-4, S-1  
(D) P-3, Q-2, R-1, S-4

Q.No. 19
During extrusion cooking, food materials are generally subjected to a combination of

(A) high shear and low pressure  
(B) high temperature and high shear  
(C) low shear and high temperature  
(D) low shear and low pressure

Q.No. 20
An orange juice flowing at 0.80 kg s\(^{-1}\) enters a counter current double pipe heat exchanger at 20 °C and leaves at 72 °C. Inlet and outlet temperatures of the hot water used as heating medium in the exchanger are 81 °C and 74 °C, respectively. The specific heat of the orange juice is 3.74 kJ kg\(^{-1}\) K\(^{-1}\) and overall heat transfer coefficient is 492 W m\(^{-2}\) K\(^{-1}\). The heat transfer surface area (round off to 2 decimal places) will be ______ m\(^2\).